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## (+)-(4R,5S)-3-[2(S)-(4-Isobutylphenyl)propionyl]-4-methyl-5-phenyloxazolidin-2-one

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## Key indicators

Single-crystal X-ray study
$T=160 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.042$
$w R$ factor $=0.101$
Data-to-parameter ratio $=8.4$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]In the title compound, $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NO}_{3}$, formed from enantiomerically pure (+)-(4R,5S)-4-methyl-5-phenyl-2-oxazolidinone and racemic 2-(4-isobutylphenyl)propanoyl chloride, the two carbonyl groups are oriented anti to each other, and the methyl group of the (4-isobutylphenyl)propionyl substituent lies close to the mean plane of the five-membered ring.

## Comment

The title compound, (I), is the second in a series of structurally related compounds, introduced in our previous report (Coumbarides et al., 2006). With $R^{1}=4-\left({ }^{i} \mathrm{Bu}\right) \mathrm{C}_{6} \mathrm{H}_{4}$, the reaction shown in that report yielded the anti-syn and syn-syn diastereomers in 34 and $32 \%$ yields, respectively. The title compound, (I), is the syn-syn diastereomer (Fig. 1). In the crystal structure, the conformation of the central portion of the molecule is closely comparable with that in the phenyl derivative (Coumbarides et al., 2006). The conformation of the five-membered ring is similar, with atoms C 1 and C 2 lying respectively 0.170 (6) $\AA$ above and 0.298 (6) $\AA$ below the plane defined by atoms O1, O2, N1 and C3. The carbonyl groups $(\mathrm{C} 3=\mathrm{O} 2$ and $\mathrm{C} 11=\mathrm{O} 3)$ are oriented anti to each other, with the torsion angle $\mathrm{O} 3-\mathrm{C} 11-\mathrm{N} 1-\mathrm{C} 3=$ $-179.7(4)^{\circ}$. The principal difference between the conformations of (I) and the phenyl derivative lies in the orientation of the (4-isobutylphenyl)propionyl substituent with respect to the remainder of the molecule: in (I), the torsion angle N1$\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13=-90.2(4)^{\circ}$ compared with $-166.97(16)^{\circ}$ for the comparable measurement in the phenyl derivative. Thus, the C19 methyl group lies closer to the plane of the fivemembered ring in (I), in contrast with the anti arrangement observed for the C4 and C19 methyl groups in the phenyl derivative.

(I)

## Experimental

The experimental procedure is comparable with that reported previously (Coumbarides et al., 2006). The actual quantities used for preparation of (I) were: $n$-butyllithium ( $0.22 \mathrm{ml}, 2.5 \mathrm{M}$ in hexanes, $0.56 \mathrm{mmol})$ and ( $R, S$ )-oxazolidinone ( $0.1 \mathrm{~g}, 0.56 \mathrm{mmol}$ ) in 10 ml tetrahydrofuran (THF), combined with a solution of (rac)-2-(4-

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Figure 1
The molecular structure of the title compound, showing the atomlabelling scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms have been omitted.
isobutylphenyl)propanoyl chloride ( $0.125 \mathrm{~g}, 0.56 \mathrm{mmol}$ ) in 1 ml THF. The crude residue was purified by flash column chromatography on silica gel, eluting with light petroleum (b.p. 313-333 K)/diethyl ether (1:1) to give a separable diastereoisomeric mixture (in the ratio anti-syn:syn-syn 54:46). The syn-syn diastereomer was obtained as colourless crystals $\left\{67 \mathrm{mg}, 32 \%\right.$ yield, m.p. 394-396 K, $R_{\mathrm{F}} 0.55$ [light petroleum (b.p 313-333 K)/diethyl ether, 1:1]\}. Spectroscopic analysis: $[\alpha]_{\mathrm{D}}^{25}=+98.1\left(\mathrm{CHCl}_{3}, 293 \mathrm{~K}\right.$, concentration 1.3 g per 100 ml$)$; IR ( $\left.\mathrm{CHCl}_{3}, \nu_{\text {max }}, \mathrm{cm}^{-1}\right): 1770(\mathrm{C}=\mathrm{O}), 1699(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right): \delta 7.38-7.16(7 \mathrm{H}, m, 7 \times \mathrm{CH} ; \mathrm{Ar}$ and Ph$), 7.08(2 \mathrm{H}$, $d, J=8.2 \mathrm{~Hz}, 2 \times \mathrm{CH} ; \mathrm{Ar}), 5.63(1 \mathrm{H}, d, J=7.4 \mathrm{~Hz}, \mathrm{CHO}), 5.05(1 \mathrm{H}, q$, $J=7.1 \mathrm{~Hz}, \mathrm{ArCH}), 4.81(1 \mathrm{H}, m, \mathrm{CHN}), 2.43\left(2 \mathrm{H}, d, J=7.2 \mathrm{~Hz}, \mathrm{CH}_{2}\right)$, 1.89-1.79 ( $\left.1 \mathrm{H}, m, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$ ), $1.48\left(3 \mathrm{H}, d, J=7.1 \mathrm{~Hz}, \mathrm{CH}_{3} \mathrm{CHAr}\right)$, $0.88\left(3 \mathrm{H}, d, J=6.7 \mathrm{~Hz}, \mathrm{CH}_{3}{ }^{a} \mathrm{CHCH}_{3}{ }^{b}\right), 0.87(3 \mathrm{H}, d, J=6.7 \mathrm{~Hz}$, $\left.\mathrm{CH}_{3}{ }^{a} \mathrm{CHCH}_{3}{ }^{b}\right), 0.72\left(3 \mathrm{H}, d, J=6.7 \mathrm{~Hz}, \mathrm{CH}_{3} \mathrm{CHN}\right) ;{ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta 174.6(\mathrm{NC}=\mathrm{O}), 152.6(\mathrm{OC}=\mathrm{O}), 140.5(i-\mathrm{C}$; Ar), 137.5 ( $i-\mathrm{C} ; \mathrm{Ar}), 133.6(i-\mathrm{C} ; \mathrm{Ph}), 129.3,127.7(2 \times \mathrm{CH} ; \mathrm{Ar}), 129.1$, 128.7 and $125.8(3 \times \mathrm{CH} ; \mathrm{Ph})$, $78.8(\mathrm{OCHPh})$, $54.7(\mathrm{CHN}), 45.1$ $\left[\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right], 42.2(\mathrm{ArCH}), 30.1 \quad\left(\mathrm{CH}_{2}\right), \quad 22.5 \quad\left(\mathrm{CH}_{3}{ }^{a} \mathrm{CHCH}_{3}{ }^{b}\right.$; isobutylphenyl), $19.4\left(\mathrm{CH}_{3} \mathrm{CH}\right)$, $14.2\left(\mathrm{CH}_{3} \mathrm{CHN}\right)$; found: $M 365.1986$; $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NO}_{3}$ requires 365.1985 .

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NO}_{3} \\
& M_{r}=365.46 \\
& \text { Orthorhombic, } P 2_{1} 2_{2} 2_{1} \\
& a=7.194(4) \AA \\
& b=14.208(8) \AA \\
& c=20.049(11) \AA \\
& V=2049(2) \AA^{3}
\end{aligned}
$$

## Data collection

Enraf-Nonius CAD-4
diffractometer
$\omega / 2 \theta$ scans
Absorption correction: none
2141 measured reflections
2079 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.042$
$w R\left(F^{2}\right)=0.101$
$S=1.00$
2079 reflections
248 parameters

1340 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.004$
$\theta_{\text {max }}=25.0^{\circ}$
2 standard reflections frequency: 60 min intensity decay: $1 \%$

H-atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0466 P)^{2}\right]$
where $P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3$
$(\Delta / \sigma)_{\max }<0.001$ 。
$\Delta \rho_{\max }=0.18 \mathrm{e}^{\circ}{ }^{-3}$
$\Delta \rho_{\text {min }}=-0.20 \mathrm{e}^{-3}$

H atoms were placed in geometrically idealised positions and constrained to ride on their parent atoms, with $\mathrm{C}-\mathrm{H}=0.95-1.00 \AA$ and $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$ or $1.5 U_{\text {eq }}$ (methyl C). The methyl groups were allowed to rotate about their local threefold axes. In the absence of significant anomalous scattering effects, Friedel pairs have been merged. The absolute configuration is assigned on the basis of the known configuration of the starting material (Coumbarides et al., 2006).

Data collection: CAD-4 EXPRESS (Enraf-Nonius, 1994); cell refinement: CAD-4 EXPRESS; data reduction: XCAD4 (Harms \& Wocadlo, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX publication routines (Farrugia, 1999).

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