Received 27 June 2006

Accepted 12 August 2006

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Gregory S. Coumbarides,^a Marco Dingjan,^a Jason Eames,^b* Majid Motevalli^a and Nela Malatesti^c

^aDepartment of Chemistry, Queen Mary, University of London, Mile End Road, London E1 4NS, England, ^bDepartment of Chemistry, University of Hull, Cottingham Road, Kingstonupon-Hull HU6 7RX, England, and ^cDepartment of Chemistry, J. J. Strossmayer University of Osijek, Trg Sv. Trojstva 3, Osijek 31000, Croatia

Correspondence e-mail: j.eames@hull.ac.uk

Key indicators

Single-crystal X-ray study T = 160 KMean σ (C–C) = 0.005 Å R factor = 0.042 wR factor = 0.101 Data-to-parameter ratio = 8.4

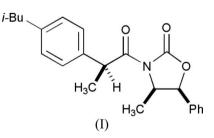
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. In the title compound, $C_{23}H_{27}NO_3$, formed from enantiomerically pure (+)-(4*R*,5*S*)-4-methyl-5-phenyl-2-oxazolidinone and racemic 2-(4-isobutylphenyl)propanoyl chloride, the two carbonyl groups are oriented *anti* to each other, and the methyl group of the (4-isobutylphenyl)propionyl substituent lies close to the mean plane of the five-membered ring.

4-methyl-5-phenyloxazolidin-2-one

(+)-(4*R*,5*S*)-3-[2(*S*)-(4-Isobutylphenyl)propionyl]-

Comment

The title compound, (I), is the second in a series of structurally related compounds, introduced in our previous report (Coumbarides *et al.*, 2006). With $R^1 = 4 - ({}^{i}Bu)C_6H_4$, the reaction shown in that report yielded the anti-syn and syn-syn diastereomers in 34 and 32% yields, respectively. The title compound, (I), is the syn-syn diastereomer (Fig. 1). In the crystal structure, the conformation of the central portion of the molecule is closely comparable with that in the phenyl derivative (Coumbarides et al., 2006). The conformation of the five-membered ring is similar, with atoms C1 and C2 lying respectively 0.170 (6) Å above and 0.298 (6) Å below the plane defined by atoms O1, O2, N1 and C3. The carbonyl groups (C3=O2 and C11=O3) are oriented anti to each other, with the torsion angle O3-C11-N1-C3 = -179.7 (4)°. The principal difference between the conformations of (I) and the phenyl derivative lies in the orientation of the (4-isobutylphenyl)propionyl substituent with respect to the remainder of the molecule: in (I), the torsion angle N1- $C11-C12-C13 = -90.2 (4)^{\circ}$ compared with $-166.97 (16)^{\circ}$ for the comparable measurement in the phenyl derivative. Thus, the C19 methyl group lies closer to the plane of the fivemembered ring in (I), in contrast with the anti arrangement observed for the C4 and C19 methyl groups in the phenyl derivative.



Experimental

The experimental procedure is comparable with that reported previously (Coumbarides *et al.*, 2006). The actual quantities used for preparation of (I) were: *n*-butyllithium (0.22 ml, 2.5 M in hexanes, 0.56 mmol) and (*R*,*S*)-oxazolidinone (0.1 g, 0.56 mmol) in 10 ml tetrahydrofuran (THF), combined with a solution of (*rac*)-2-(4-

© 2006 International Union of Crystallography All rights reserved

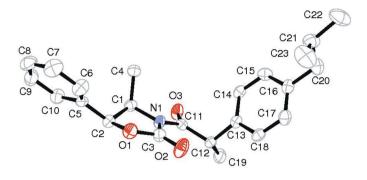


Figure 1

The molecular structure of the title compound, showing the atomlabelling scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms have been omitted.

isobutylphenyl)propanoyl chloride (0.125 g, 0.56 mmol) in 1 ml THF. The crude residue was purified by flash column chromatography on silica gel, eluting with light petroleum (b.p. 313-333 K)/diethyl ether (1:1) to give a separable diastereoisomeric mixture (in the ratio antisyn:syn-syn 54:46). The syn-syn diastereomer was obtained as colourless crystals {67 mg, 32% yield, m.p. 394-396 K, R_F 0.55 [light petroleum (b.p 313-333 K)/diethyl ether, 1:1]}. Spectroscopic analysis: $\left[\alpha\right]_{D}^{25} = +98.1$ (CHCl₃, 293 K, concentration 1.3 g per 100 ml); IR (CHCl₃, ν_{max} , cm⁻¹): 1770 (C=O), 1699 (C=O); ¹H NMR (250 MHz; CDCl₃): δ 7.38–7.16 (7H, m, 7 × CH; Ar and Ph), 7.08 (2H, d, J = 8.2 Hz, $2 \times$ CH; Ar), 5.63 (1H, d, J = 7.4 Hz, CHO), 5.05 (1H, q, q) J = 7.1 Hz, ArCH), 4.81 (1H, m, CHN), 2.43 (2H, d, J = 7.2 Hz, CH₂), 1.89–1.79 (1H, m, CH(CH₃)₂)), 1.48 (3H, d, J = 7.1 Hz, CH₃CHAr), 0.88 (3H, d, J = 6.7 Hz, CH₃^{*a*}CHCH₃^{*b*}), 0.87 (3H, d, J = 6.7 Hz, $CH_3^{a}CHCH_3^{b}$), 0.72 (3H, d, J = 6.7 Hz, CH_3CHN); ¹³C NMR (100.6 MHz; CDCl₃): § 174.6 (NC=O), 152.6 (OC=O), 140.5 (i-C; Ar), 137.5 (i-C; Ar), 133.6 (i-C; Ph), 129.3, 127.7 (2 × CH; Ar), 129.1, 128.7 and 125.8 (3 × CH; Ph), 78.8 (OCHPh), 54.7 (CHN), 45.1 $[CH(CH_3)_2], 42.2$ (ArCH), 30.1 (CH₂), 22.5 (CH₃^{*a*}CHCH₃^{*b*}; isobutylphenyl), 19.4 (CH₃CH), 14.2 (CH₃CHN); found: M 365.1986; C₂₃H₂₇NO₃ requires 365.1985.

Crystal data

 $\begin{array}{l} C_{23}H_{27}NO_{3}\\ M_{r}=365.46\\ \text{Orthorhombic, }P2_{1}2_{1}2_{1}\\ a=7.194 \ (4) \ \text{\AA}\\ b=14.208 \ (8) \ \text{\AA}\\ c=20.049 \ (11) \ \text{\AA}\\ V=2049 \ (2) \ \text{\AA}^{3} \end{array}$

Z = 4 $D_x = 1.185 \text{ Mg m}^{-3}$ Mo K\alpha radiation $\mu = 0.08 \text{ mm}^{-1}$ T = 160 (2) KBlock, colourless $0.63 \times 0.38 \times 0.15 \text{ mm}$

Data collection

Enraf–Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: none 2141 measured reflections 2079 independent reflections

RefinementRefinement on F^2 H-atom $R[F^2 > 2\sigma(F^2)] = 0.042$ $w = 1/[\sigma]$ $wR(F^2) = 0.101$ whereS = 1.00 $(\Delta/\sigma)_{max}$ 2079 reflections $\Delta\rho_{max}$ 248 parameters $\Delta\rho_{min}$

1340 reflections with $I > 2\sigma(I)$ $R_{int} = 0.004$ $\theta_{max} = 25.0^{\circ}$ 2 standard reflections frequency: 60 min intensity decay: 1%

H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0466P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.18 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.20 \text{ e} \text{ Å}^{-3}$

H atoms were placed in geometrically idealised positions and constrained to ride on their parent atoms, with C-H = 0.95-1.00 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ or $1.5U_{eq}(methyl C)$. The methyl groups were allowed to rotate about their local threefold axes. In the absence of significant anomalous scattering effects, Friedel pairs have been merged. The absolute configuration is assigned on the basis of the known configuration of the starting material (Coumbarides *et al.*, 2006).

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* publication routines (Farrugia, 1999).

We are grateful to Onyx Scientific Limited (Drs Tony Flinn and Julian Northen) and Queen Mary, University of London for a studentship to MD, the Royal Society and the University of London Central Research Fund for financial support to JE, and the EPSRC National Mass Spectrometry Service (Swansea) for accurate mass determination.

References

Coumbarides, G. S., Eames, J., Motevalli, M., Malatesti, N. & Yohannes, Y. (2006). Acta Cryst. E62, 04032–04034.

Enraf-Nonius (1994). CAD-4 EXPRESS. Enraf-Nonius, Delft, The Netherlands.

- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.